

## Science AQA GCSE Combined and GCSE Biology, Chemistry and Physics

	GCSE Unit, Topic or Summary of work covered (NB Order may differ depending on specialist teacher pairing.)	Knowledge & Skills Developed	Assessment	Personal Development
Autumn 1	Physics: Electricity and Particle Model of Matter,	Knowledge: Current, potential difference and resistance Series and parallel circuits Domestic uses and safety Energy transfers Changes of state and the particle model Internal energy and energy transfers Particle model and pressure Working Scientifically:	Electricity Exam Assessed Required Practicals	Safe use of domestic electricity Energy Efficiency
		Development of Scientific thinking, Analysis and Evaluation, Scientific vocabulary <b>Maths Skills:</b> Arithmetic and numerical computation, Handling Data, Algebra, Graphs		
Autumn 2	Physics: Atomic Structure, Forces 1	Knowledge: Atoms and isotopes Atoms and nuclear radiation Forces and their interactions Work done and energy transfer Forces and elasticity Forces and motion Speed/Velocity Newton's Laws Stopping Distances Momentum (HT only) Working Scientifically: Development of Scientific Thinking, Analysis and Evaluation, Scientific vocabulary, Experimental Skills and Strategies	Particle Model of Matter Exam Assessed Required Practicals	Stopping distances

		Maths Skills: Arithmetic and numerical computation,		
		Handling Data, Algebra, Graphs		
Spring 1	Biology: Organisation (2), Homeostasis,	Knowledge: Homeostasis The human nervous system Human endocrine system Control of blood glucose concentration Hormones in human reproduction Contraception The use of hormones to treat infertility (HT only) Negative feedback (HT only)	Assessed Required Practicals	Contraception Fertility Exercise Diabetes
		Working Scientifically: Development of Scientific Thinking Maths Skills: Arithmetic and numerical computation, Handling Data, Algebra, Graphs		
Spring 2	Biology: Infection and Response, Bioenergetics	Knowledge:Communicable (infectious) diseasesViral diseasesBacterial diseasesFungal diseasesFungal diseasesProtist diseasesHuman defence systemsVaccinationAntibiotics and painkillersDiscovery and development of drugsPhotosynthetic reactionRate of photosynthesisUses of glucose from photosynthesisAerobic and anaerobic respirationResponse to exerciseMetabolismWorking Scientifically: Development of ScientificThinkingMaths Skills: Arithmetic and numerical computation,Handling Data, Algebra, Graphs	Bioenergetics Exam Infection and Response Exam Assessed Required Practicals	Hygiene Infection Control (vaccination) STIs Legal and Illegal drugs

Chemistry: Bonding and Structure, Chemical Changes (2)	Knowledge:Conservation of mass and balanced chemicalequationsRelative formula massMass changes when a reactant or product is a gasChemical measurementsMoles (HT only)Amounts of substances in equations (HT only)Using moles to balance equations (HT only)Limiting reactants (HT only)Concentration of solutionsMetal oxidesThe reactivity seriesExtraction of metals and reductionOxidation and reduction in terms of electrons (HTonly)Reactions of acids with metalsNeutralisation of acids and salt productionSoluble saltsThe pH scale and neutralisationStrong and weak acids (HT only)The process of electrolysisElectrolysis of molten ionic compoundsUsing electrolysis to extract metalsElectrolysis of aqueous solutionsRepresentation of reactions at electrodes as halfequations (HT only)Working Scientifically: Development of ScientificThinking , Analysis and Evaluation, ScientificThinking , Analysis and Evaluation, ScientificNumber Skills: Arithmetic and numerical computation,Hand Miter Data	Chemical Changes Exam Assessed Required Practicals	
Chemistry: Quantitative Chemistry, Energy Changes	Handling Data, Algebra   Knowledge:   Energy transfer during exothermic and endothermic reactions   Reaction profiles	Quantitative Chemistry Exam Assessed Required Practicals	Energy consumption and links between CO2 and climate change
	Structure, Chemical Changes (2)	Structure, Chemical Changes (2)Conservation of mass and balanced chemical equations Relative formula mass Mass changes when a reactant or product is a gas Chemical measurements Moles (HT only) Amounts of substances in equations (HT only) Using moles to balance equations of solutions Metal oxides The reactivity series Extraction of metals and reduction Oxidation and reduction in terms of electrons (HT only) Reactions of acids and salt production Soluble salts 	Structure, Chemical Changes Conservation of mass and balanced chemical equations Chemical Changes Exam   (2) Relative formula mass Assessed Required Practicals   Relative formula mass Mass changes when a reactant or product is a gas Chemical measurements   Moles (HT only) Amounts of substances in equations (HT only) Limiting reactants (HT only)   Concentration of solutions Metal oxides The reactivity series   Extraction of metals and reduction Oxidation and reduction in terms of electrons (HT only) Representation of acids and salt production   Soluble salts The pH scale and neutralisation Strong and weak acids (HT only) Representation of reactions of moles as half equations (HT only)   Using electrolysis of molten ionic compounds Using electrolysis of molten ionic compounds Using electrolysis of molten ionic compounds   Using electrolysis of aqueous solutions Representation of reactions at electrodes as half equations (HT only) Working Scientifically: Development of Scientific   Working Scientifically: Arithmetic and numerical computation, Handling Data, Algebra Quantitative Chemistry Exam   Chemistry: Quantitative Knowlege: Energy transfer during exothermic and endothermic reactions

Calculating rates of reactions Factors which affect the rates of chemical reactions Collision theory and activation energy Catalysts Reversible reactions Energy changes and reversible reactions Equilibrium The effect of changing conditions on equilibrium (HT
only) The effect of changing concentration (HT only) The effect of temperature changes on equilibrium (HT only) The effect of pressure changes on equilibrium (HT only)
Working Scientifically: Development of Scientific   Thinking,   Maths Skills: Arithmetic and numerical computation,   Algebra, Graphs and Geometry



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Autumn 1	Physics: Forces 2, Waves	Knowledge: Ways of reducing the use of resources Waves in air, fluids and solids Electromagnetic waves Uses and applications of Electromagnetic Waves Working Scientifically: Development of Scientific Thinking, Analysis and Evaluation, Scientific vocabulary, Experimental Skills and Strategies Maths Skills: Arithmetic and numerical computation, Handling Data, Algebra, Graphs	Waves Exam Assessed Required Practicals	Sustainability – use of resources
Autumn 2	Physics: Magnetism and Electromagnetism	Knowledge: Permanent and induced magnetism Poles of a Magnet Magnetic forces Magnetic Fields The motor effect Fleming's Left-hand rule (HT Only) Working Scientifically: Analysis and Evaluation, Scientific vocabulary, Experimental Skills and Strategies Maths Skills: Arithmetic and numerical computation, Handling Data, Algebra, Graphs	Assessed Required Practicals	
Spring 1	Biology: Inheritance, Variation and Evolution and Ecology	Knowledge: Sexual and asexual reproduction Meiosis DNA and the genome Genetic inheritance Inherited disorders	Inheritance, Variation and Evolution Exam Assessed Required Practicals	

Spring 2	2 Chemistry: Rate and Extent of Chemical Changes, Organic Chemistry, Chemical Analysis, Chemistry of the Atmosphere, Using Resources	Sex determination Working Scientifically: Development of Scientific Thinking, Experimental Skills and Strategies Maths Skills: Arithmetic and numerical computation, Handling Data, Algebra, Graphs Knowledge: Pure substances Formulations Chromatography Identification of common gases The proportions of different gases in the atmosphere The Earth's early atmosphere How oxygen increased How carbon dioxide decreased Greenhouse gases Human activities which contribute to an increase in greenhouse gases in the atmosphere Global climate change The carbon footprint and its reduction Atmospheric pollutants from fuels Properties and effects of atmospheric pollutants	Organic Chemistry Exam Assessed Required Practicals Y11 Mocks (Paper 1 B, C, P)	Use of finite natural resources Pollution Climate Change Sustainability
		Using the Earth's resources and sustainable development Potable water Waste water treatment Alternative methods of extracting metals (HT only) Life cycle assessment Crude oil, hydrocarbons and alkanes Fractional distillation and petrochemicals Properties of hydrocarbons Cracking and alkenes <b>Working Scientifically:</b> Development of Scientific Thinking, Analysis and Evaluation, Scientific vocabulary, Experimental Skills and Strategies		

		Maths Skills: Arithmetic and numerical computation, Handling Data, Algebra, Graphs		
Summer 1	Revision and exam practice		Practice Exam Papers Required Practical Questions 6 Mark Question Practice	
Summer 2	Revision and exam practice		Practice Exam Papers Assessed Required Practicals 6 Mark Question Practice	